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Highly diastereo-/enantioselective Cu-catalyzed propargylic alkylations of propargyl acetates with cyclic enamines†

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Highly diastereo- and enantioselective copper-catalyzed propargylic alkylations of propargylic acetates with morpholine-derived cyclic enamines for the construction of vicinal tertiary stereocenters have been developed. By the employment of a less sterically hindered chiral tridentate P,N,N-ligand, good to excellent diastereo- (up to >98 : 2 dr) and enantioselectivity (up to 99% ee) could be achieved for a wide range of substrates.

Transition-metal-catalyzed propargylic substitutions represent an important class of reactions due to their ability to introduce an electron-rich triple bond that is a versatile entity for further chemical transformations.¹ In 2008, van Maarseveen² and Nishibayashi³ independently reported the first copper-catalyzed asymmetric propargylic amination. Since then, a variety of nitrogen, oxygen, and carbon nucleophiles have proved to be suitable reagents for the copper-catalyzed asymmetric propargylic substitution.⁴ However, the methods for Cu-catalyzed propargylic substitution that employ prochiral nucleophiles to build two vicinal stereocenters and display high diastereo- and enantioselectivity remain elusive.⁵

Recently, our group have demonstrated the power of copper catalysis in accessing vicinal tertiary stereocenters with our report on the diastereo- and enantioselective asymmetric propargylic alkylation of acyclic enamines (Scheme 1A).⁶ However, the use of cyclic enamines was less successful, leading to poor chemoselectivity. The success of this protocol combined

with the virtual absence of successful reports describing the application of this transformation to cyclic enamines⁷ encouraged our further exploration of copper catalysts in the domain of this substrate class. In fact, our previous studies have disclosed that a highly diastereo- and enantioselective [3 + 3] cycloaddition instead of propargylic alkylation preferentially took place for the reaction between propargylic esters and cyclic enamines although this cycloaddition is believed to proceed *via* the propargylic alkylation as the key step (Scheme 1B).⁸ It's therefore anticipated that an enantioselective propargylic alkylation of propargylic esters with cyclic enamines could be realized if the last cyclization step can be efficiently inhibited. Herein we report a highly diastereo- and enantioselective



B. Previous work on [3+3] cycloaddition with cyclic enamines (ref. 8):



C. Current work on propargylic alkylation of cyclic enamines:



Scheme 1 Copper-catalyzed asymmetric reaction between propargylic esters with enamines.

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copper-catalyzed propargylic alkylation of propargylic acetates with morpholine-derived cyclic enamines by the employment of a less sterically hindered chiral tridentate P,N,N-ligand to build up two vicinal tertiary stereocenters, in which >98 : 2 dr and up to 99% ee for *syn*-diastereoisomers have been achieved (Scheme 1C).

Considering the significant influence of the amino moiety of enamine substrates on the outcomes in the copper-catalyzed [3 + 3] cycloaddition,⁸ our initial attempt focused on probing the effect of enamines on the reaction. As shown in Table 1, the amino moiety of enamines displayed the crucial role in the chemoselectivity of the reaction. When *N*,*N*-diethylamino substituted enamine **2a** was used as the reaction partner, only [3 + 3] cycloadduct was obtained and no alkylation product was detected (entry 1). By the introduction of cyclic amino group into enamines, the formation of the alkylation product **3a** was observed (entries 2–4). In particular, the use of enamine **2d** bearing a morpholine structure as the nucleophile predominantly gave rise to the alkylation product **3a** in 80% yield (entry 4).

Having established an efficient copper-catalyzed propargylic alkylation of cyclic enamine 2d, we next investigated the effects of different ligands, copper salts, reaction temperature on the efficiency and selectivity of the reaction. 1-Phenyl-2-propynyl acetate 1a and 4-(cyclohex-1-en-1-yl)morpholine 2d were chosen as standard reaction partners, and some selected results of these experiments are summarized in Table 2. Initially, the effect of the ligand was investigated (Fig. 1). As we have disclosed, the use of chiral ferrocenyl P,N,N-ligand L₂ led to the predominant formation of the alkylation products 3a with *syn*-3a as the major diastereoisomer in 91% ee (entry 2). An attempt to improve the reaction outcome by the use of structurally rigid and bulky ketimine P,N,N-ligand L₁, which displayed excellent

Table 1 Screening the reaction between propargylic acetate 1a and enamines $\mathbf{2}^a$

OA Ph	Ac + (R_c, i_p)	Cu(OAc) ₂ H ₂ O (5 mol%) S _p)-L ₂ (5.5 mol%) r_2 NEt (1.2 equiv) MeOH, rt, 12 h (syn + an	b)-3a 4
	NEt ₂ N		
Entry	Enamine (2)	$3a (syn/anti) : 4^b$	Yield of $3a^{c}$ (%)
1	2a	<2 ():>98	_
2	2b	22 (71/29) : 78	16
3	2 c	36 (78/22): 64	27
4	2d	94 (91/9) : 6	80

^{*a*} Reaction conditions: Cu(OAc)₂·H₂O (0.015 mmol), (R_c , S_p)-L₂ (0.0165 mmol), **1a** (0.3 mmol), **2** (0.36 mmol), ^{*i*}Pr₂NEt (0.36 mmol) were stirred in 2 mL of MeOH at rt for 12 h. ^{*b*} Determined by ¹H NMR. ^{*c*} Isolated yield after column chromatography.

diastereo-/enantioselectivity in the Cu-catalyzed asymmetric propargylic alkylation of acyclic enamines,6 however, was disappointed. Only poor chemoselectivity (3a: 4 = 61: 39) was obtained although the enantioselectivity for syn-3a was high (entry 1). This should be high reactivity of copper/ L_1 complex that couldn't efficiently suppress the last cyclization step. We therefore envisioned that ligand having lower reactivity should be more suitable to this reaction. A less sterically hindered P,N,N-ligand L₃ was therefore employed in the reaction, and as expected, gave the similar result to that with L₂ (entry 3). Excellent chemo- and diasteroselectivity was observed with (S,S)-Ph-pybox L₄ as ligand, however, enantioselectivity for syn-3a was low (entry 4). With (S)-SegPhos L_5 , high chemoselectivity but low diasteteo- and enantioselectivity was achieved (entry 5). Following investigation of copper salts didn't improve the reaction outcome (entries 6-8). Lowering the reaction temperature to 0 °C could significantly improve the chemoselectivity, exclusively giving the alkylation product 3a in good diastereoselectivity, regardless of the use of L₂ or L₃ (entries 9–10). Especially, ligand L_3 led to the alkylation product 3a in 82% yield, and with 94/6 dr and an ee-value of up to 94% for the major syn-diastereoisomer (entry 10).

Having identified the optimal set of reaction conditions, we then investigated the scope of this reaction with respect to propargylic acetates, and the results are shown in Table 3. We found that the reaction proceeded smoothly for all 1-aryl-2propynyl acetates. In all cases, only alkylation products, predominately syn-diastereoisomers, were obtained and no [3 + 3] cycloadduct was detected. The position of the substituent on the phenyl ring had less influence on the reaction. Thus, all three substrates bearing a Cl-group at the different position on the phenyl ring gave the similarly good results (entries 2-4). The electronic properties of the substituent at the para-position showed little effect on the diastereo-/enantioselectivity. Propargylic acetates bearing either electron-donating or electronwithdrawing substituents on the phenyl ring fared very well in the reaction, delivering the alkylation products in high yield, diastereo-, and enantioselectivity (entries 5-8). 2-Naphthyl substrate 1i also worked well, gave the alkylation product 3i in 88% yield, 94:6 dr and 92% ee for syn-diastereoisomer (entry 9). We were pleased to discover that the use of heteroarylsubstituted propargylic acetates (substrates 1j and 1k) resulted in smooth reactions and delivered the alkylation products 3j and 3k with high yield and diastereoselectivity and with good to excellent enantioselectivity (entries 10-11).

To further evaluate this asymmetric protocol, we next examined the diversity of cyclic enamines permitted in the reaction, and the results are listed in Table 4. In all cases, the reaction proceeded smoothly to afford the alkylation products as the only outcome with *syn*-diastereoisomers as the major products. In comparison to 4-(cyclohex-1-en-1-yl)morpholine **2d**, five-membered cyclic enamines gave the similar yield and diastereoselectivity, but a slightly decreased enantioselectivity. The substituents in cyclohexenyl ring were well tolerated, delivering the corresponding alkylation products in high yield and with excellent diastereo- and enantioselectivity. Importantly, aliphatic acyclic ketone enamine also served well as the



Entry	[Cu]	L*	syn-3a : anti-3a : 4^b	Yield of $3a^{c}$ (%)	ee of syn- $3a^{d}$ (%)
1	$Cu(OAc)_2 \cdot H_2O$	L1	55:6:39	51	94
2	$Cu(OAc)_2 \cdot H_2O$	L2	86:8:6	80	91
3	$Cu(OAc)_2 \cdot H_2O$	L3	84:8:8	76	92
4	$Cu(OAc)_2 \cdot H_2O$	L4	97:3:0	88	42
5	$Cu(OAc)_2 \cdot H_2O$	L5	55:45:0	63	$52(51)^{f}$
6	CuI	L3	80:20:0	74	76 $(56)^{f}$
7	$CuOTf \cdot (C_6H_6)_{1/2}$	L3	89:7:4	72	92
8	$Cu(OTf)_2$	L3	88:7:5	72	90
9^e	$Cu(OAc)_2 \cdot H_2O$	L2	90:10:0	67	92
10^e	$Cu(OAc)_2 \cdot H_2O$	L3	94:6:0	82	94

^{*a*} Reaction conditions: copper salt (0.015 mmol), **L*** (0.0165 mmol), **1a** (0.3 mmol), **2d** (0.36 mmol), ^{*i*}Pr₂NEt (0.36 mmol) were stirred in 2 mL of MeOH at rt for 12 h unless otherwise specified. ^{*b*} Determined by GC or ¹H NMR. ^{*c*} Isolated yield of *syn*-**3a** and *anti*-**3a** in total. ^{*d*} Determined by HPLC on chiral column. ^{*e*} The reaction was performed at 0 °C for 10 h. ^{*f*} The ee-values of *anti*-**3a** in parentheses.



Fig. 1 Ligand evaluated in the Cu-catalyzed asymmetric propargylic alkylation.

reaction partner although the use of (S)-L₁ instead of (R)-L₃ was required. In this case, (S)-L₁ showed better performance than (R)-L₃ presumably due to the steric effect, giving the alkylation product 3**q** in moderate yield and diastereoselectivity but with perfect ee of 99%.

The absolute configuration of the propargylic alkylation product was unambiguously determined by X-ray structure analysis of **3r**,⁹ which was prepared by the reaction of 1-(4bromophenyl)prop-2-yn-1-yl acetate **1f** with 4-(1,4-dioxaspiro [4.5]dec-7-en-8-yl)morpholine in the catalysis of Cu(OAc)₂·H₂O and (*S*)-L₃. A (*S*,*R*)-absolute configuration was assigned as shown in Scheme 2, which is corresponding to *syn*diastereoisomer.

Based on the experimental results and previous reports,^{4b,7b} a reaction pathway is proposed as shown in Scheme 3. In the

first step, a copper complex forms the π -complex **A** with propargylic acetate **1a**. Deprotonation of **A** with a base then gives a Cu-acetylide complex **B**, which would explain why

Table 3Scope of propargylic esters $\mathbf{1}^a$



Entry	1 (R)	3	Yield of 3^{b} (%)	dr ^c	ee of <i>syn-3^d</i> (%)
1	1a: R = Ph	3a	82	94:6	94
2	1b: $R = 4$ -ClC ₆ H ₄	3b	80	93:7	95
3	1c: $R = 3-ClC_6H_4$	3c	85	>98:2	97
4	1d: $R = 2 - ClC_6H_4$	3d	88	>98:2	95
5	1e: $R = 4$ -FC ₆ H ₄	3e	88	97:3	92
6	1f: $R = 4$ -BrC ₆ H ₄	3f	90	94:6	95
7	1g: $R = 4$ - $CF_3C_6H_4$	3g	84	94:6	95
8	1h : $R = 4$ -MeC ₆ H ₄	3h	92	92:8	91
9	1i: R = 2-naphthyl	3i	88	94:6	92
10	1j : $R = 2$ -furyl	3j	90	94:6	80
11	1k: $R = 3$ -pyridyl	3k	86	91:9	95

^{*a*} Reaction conditions: Cu(OAc)₂·H₂O (0.015 mmol), L₃ (0.0165 mmol), 1 (0.3 mmol), 2d (0.36 mmol), ⁱPr₂NEt (0.36 mmol) were stirred in 2 mL of MeOH at 0 °C for 10 h. ^{*b*} Isolated yield of *syn*-3a and *anti*-3a in total. ^{*c*} Determined by GC or ¹H NMR. ^{*d*} Determined by HPLC on chiral column.

 Table 4
 Scope of morpholine-derived cyclic enamines^a



^{*a*} Reaction conditions: Cu(OAc)₂·H₂O (0.015 mmol), L₃ (0.0165 mmol), **1a** (0.3 mmol), **2d-j** (0.36 mmol), ⁱPr₂NEt (0.36 mmol) were stirred in 2 mL of MeOH at 0 °C for 10 h. ^{*b*} Isolated yield of *syn-***3a** and *anti-***3a** in total. ^{*c*} Determined by GC or ¹H NMR. ^{*d*} Determined by HPLC on chiral column. ^{*e*} (*S*)-L₁ was used instead of (*R*)-L₃.

a propargylic acetate bearing an internal alkyne moiety such as 1,3-diphenyl-2-propynyl acetate 5 did not react at all. Elimination of an acetoxyl moiety from **B** affords the copper–allenylidene complex **C** or its resonance structure, the copper–acetylide complex **D** bearing a cationic γ -carbon. The nucleophilic attack of β -carbon atom of enamine 2d to the C $_{\gamma}$ -atom of the copper–allenylidene complex (C) gives the corresponding copper–acetylide complex (E), which should be the key step for the stereoselection. The intramolecular shift of H atom generates the copper π -alkyne complex **F**. The starting complex **A** is then re-



Scheme 3 Proposed mechanism.

generated from F by liberating the propargylic alkylation product G by the ligand exchange with another propargylic acetate 1a.

A transition state of Cu–allenylidene complex with chiral P,N,N-ligand is also proposed to explain the observed stereochemistry as shown in Scheme 4. The edge-to-face aromatic interaction^{4b} and the steric hinderance make the attack of



Scheme 2 Synthesis of (*S*,*R*)-**3***r* and its absolute configuration determined by X-ray analysis.



Scheme 4 Proposed model for the diastereo-/enantioselectivities.

enamine C_{β} -nucleophile at the propargylic cation favourably from the R_e face *via* **M-2** mode to generate (*R*,*S*)-**3a**.

Conclusions

In conclusion, we have developed a highly diastereo- and enantioselective copper-catalyzed propargylic alkylation of propargylic acetates with morpholine-derived cyclic enamines for the construction of two vicinal tertiary stereocenters. The reaction was preferential to the formation of syn-diastereoisomers of the propargylic alkylation products in the complete chemoselectivity. By the employment of a chiral 1phenylethylamine-derived tridentate P,N,N-ligand, good to excellent diastereoselectivity (>98:2) and enantioselectivity (up to 99% ee) have been achieved for a wide range of substrates. In comparison with the reaction of acyclic enamines, less sterically hindered P,N,N-ligand showed better performance in the propargylic alkylation of cyclic enamines in term of chemoselectivity, suggesting that its lower reactivity efficiently supressed the last cyclization step. Further development and application of this reaction are underway.

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